

- Hoekstra, W. G. (1974). *Trace Element Metabolism in Animals 2*, edited by W. G. Hoekstra, J. W. Suttie, H. E. Ganther & W. Mertz. Baltimore: University Park Press.
- Iwaoka, M. & Tomoda, S. (1994). *J. Am. Chem. Soc.* **116**, 2557–2561.
- Jacquemin, P. V., Christiaens, L. E., Renson, M. J., Evers, M. J. & Dereu, N. (1992). *Tetrahedron Lett.* **33**, 3863–3866.
- Johnson, C. K. (1976). *ORTEP*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Molecular Structure Corporation (1994). *TEXSAN. Single Crystal Structure Analysis Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Reich, H. J. & Jasperse, C. P. (1987). *J. Am. Chem. Soc.* **109**, 5549–5551.
- Schagen, J. D., Straver, L., van Meurs, F. & Williams, G. (1989). *CAD-4 Software*. Delft Instruments X-ray Diffraction, PO Box 811, 2600 AV Delft, The Netherlands.
- Shamberger, R. J. (1983). In *Biochemistry of Selenium*. New York: Plenum Press.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Tappel, A. L. (1984). *Curr. Top. Cell Regul.* **24**, 87–97.
- Vessman, K., Ekström, M., Berglund, M., Andersson, C. M. & Engman, L. (1995). *J. Org. Chem.* **60**, 4461–4467.
- Wilson, S. R., Zucker, P. A., Huang, R.-R. C. & Spector, A. (1989). *J. Am. Chem. Soc.* **111**, 5936–5939.

Acta Cryst. (1998). **C54**, 427–428

1-Acetyl-2-thiohydantoin

JOSÉ S. CASAS,^a ALFONSO CASTIÑEIRAS,^a DELFINA COUCE,^b
NURIA PLAYÁ,^a JOSÉ SORDO^a AND JOSÉ M. VARELA^a

^aUniversidad de Santiago de Compostela, Departamento de Química Inorgánica, Facultad de Farmacia, Campus Universitario Sur, E-15706 Santiago de Compostela, Spain, and ^bUniversidad de Vigo, Departamento de Química Inorgánica, Lagoas – Marcosende, E-36200 Vigo, Spain.
E-mail: qiac01@usc.es

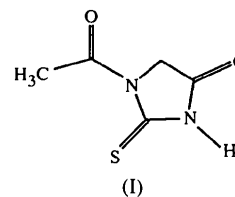
(Received 21 May 1997; accepted 31 October 1997)

Abstract

In the title compound (1-acetyl-4-oxoimidazolidine-2-thione, C₅H₆N₂O₂S), the plane of the acetyl group forms an angle of 6.7° with the essentially planar thiohydantoin ring. N—H···O hydrogen bonds create quasi-planar chains of molecules along the *y* axis.

Comment

The structure of the title compound, (I), has been established as part of a study of the synthesis and characterization of metal complexes of 2-thiohydantoin and its derivatives (Casas *et al.*, 1995).



The molecular structure of (I) is shown in Fig. 1. The N1—C1—N2—C3—C2 ring and the peripheral S, C4 and O3 atoms define a plane (r.m.s. deviation 0.013 Å), as does the *N*-acetyl N1—C4(=O4)—C5 fragment (r.m.s. deviation 0.006 Å). The angle between the two planes is 6.7°.

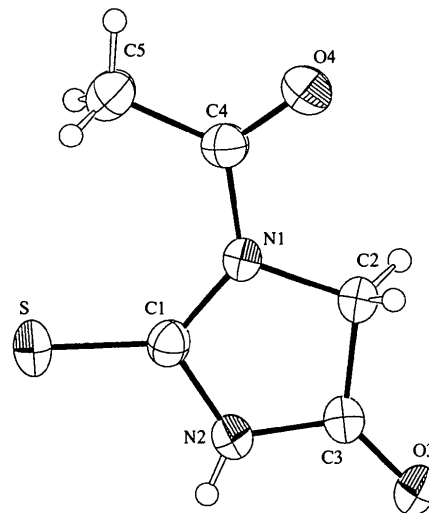


Fig. 1. The molecular structure of the title compound showing the atom-labelling scheme and 50% probability displacement ellipsoids.

The bond lengths and angles in the acetyl fragment are similar to those found in 1-acetyl-2-[1-(acetylthio)ethyl]thiohydantoin (MacKay *et al.*, 1992), although in this C2-substituted thiohydantoin, the angle between the thiohydantoin ring and the acetyl group is 12°. In (I), the thiohydantoin ring bond lengths differing most from those found in 2-thiohydantoin (Devillanova *et al.*, 1987; Walker *et al.*, 1969) are those of C1—N1 and N1—C2, which are longer in the acetyl derivative. The internal ring angle most affected by *N*-acetylation is C2—N1—C1, which widens slightly to approximately the same value as in 1-acetyl-2-[1-(acetylthio)ethyl]thiohydantoin (MacKay *et al.*, 1992). *N*-Acetylation also affects the external angles flanking the C=S group, with N1—C1—S becoming wider and N2—C1—S narrower.

The N2—H2 bond and the O3 atom are involved in a hydrogen bond [N2—H2 0.77 (3), H2···O3ⁱ 2.08 (4), N2···O3ⁱ 2.849 (3) Å and N2—H2···O3ⁱ 170 (3)°; sym-

metry code: (i) $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$] which links the molecules in quasi-planar chains along the y axis. Thus, N-acetylation simplifies the hydrogen-bond network found in 2-thiohydantoin which creates two-dimensional sheets rather than ribbons (Walker *et al.*, 1969).

Experimental

The title compound was prepared by refluxing 3.32 g of thiohydantoin in 20 ml of acetic anhydride for 30 min. The solid formed was filtered off and washed with ethyl ether. Crystals were obtained by slow evaporation of an acetone solution.

Crystal data

C₅H₆N₂O₂S
M_r = 158.18
 Monoclinic
*P*2₁/*n*
a = 8.2968 (11) Å
b = 7.7364 (11) Å
c = 10.6066 (15) Å
 β = 93.434 (11)°
V = 679.6 (2) Å³
Z = 4
D_s = 1.546 Mg m⁻³
D_m not measured

Mo *K*α radiation
 λ = 0.71073 Å
 Cell parameters from 25 reflections
 θ = 9.07–18.41°
 μ = 0.411 mm⁻¹
T = 293 (2) K
 Prism
 0.20 × 0.15 × 0.15 mm
 Colourless

Data collection

Enraf–Nonius MACH3 diffractometer
 ω scans
 Absorption correction: none
 1458 measured reflections
 1384 independent reflections
 1026 reflections with *I* > 2σ(*I*)

*R*_{int} = 0.031
 θ_{max} = 26.29°
 $h = -10 \rightarrow 10$
 $k = -9 \rightarrow 0$
 $l = 0 \rightarrow 13$
 3 standard reflections
 frequency: 120 min
 intensity decay: none

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.040
 $wR(F^2) = 0.115$
S = 1.038
 1384 reflections
 116 parameters
 H atoms refined isotropically
 $w = 1/[\sigma^2(F_o^2) + (0.0632P)^2 + 0.2702P]$
 where $P = (F_o^2 + 2F_c^2)/3$

(Δ/σ)_{max} < 0.001
 $\Delta\rho_{\text{max}} = 0.214 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\text{min}} = -0.230 \text{ e } \text{Å}^{-3}$
 Extinction correction: SHELXL93
 Extinction coefficient: 0.035 (5)
 Scattering factors from International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

S—C1	1.638 (2)	N1—C2	1.469 (3)
O3—C3	1.218 (3)	N2—C3	1.362 (3)
O4—C4	1.208 (3)	N2—C1	1.381 (3)
N1—C1	1.370 (3)	C2—C3	1.492 (3)
N1—C4	1.409 (3)	C4—C5	1.486 (4)
C1—N1—C4	131.4 (2)	N1—C2—C3	102.5 (2)
C1—N1—C2	111.3 (2)	O3—C3—N2	125.4 (2)
C4—N1—C2	117.4 (2)	O3—C3—C2	128.2 (2)
C3—N2—C1	113.6 (2)	N2—C3—C2	106.4 (2)
N1—C1—N2	106.1 (2)	O4—C4—N1	117.3 (2)
N1—C1—S	132.0 (2)	O4—C4—C5	122.3 (2)
N2—C1—S	121.9 (2)	N1—C4—C5	120.3 (2)

Data collection: CAD-4 EXPRESS Software (Enraf–Nonius, 1995). Cell refinement: CAD-4 EXPRESS Software. Data reduction: HELENA (Spek, 1996). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ZORTEP (Zsolnai, 1996). Software used to prepare material for publication: SHELXL93.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: AB1500). Services for accessing these data are described at the back of the journal.

References

- Casas, J. S., Castellano, E. E., Macias, A., Playa, N., Sánchez, A., Sordo, J., Varela, J. M. & Zukerman-Schpector, J. (1995). *Inorg. Chim. Acta*, **238**, 129–137.
 Devillanova, F. A., Isaia, F., Verani, G., Battaglia, L. P. & Corradi, A. B. (1987). *J. Chem. Res. (M)*, pp. 1617–1638.
 Enraf–Nonius (1995). CAD-4 EXPRESS Software. Version 5.1. Enraf–Nonius, Delft, The Netherlands.
 MacKay, M. F., Duggan, B. M., Laslett, R. L. & Wilshire, J. F. K. (1992). *Acta Cryst. C* **48**, 334–336.
 Sheldrick, G. M. (1990). *Acta Cryst. A* **46**, 467–473.
 Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.
 Spek, A. L. (1996). HELENA. Program for Data Reduction. University of Utrecht, The Netherlands.
 Walker, L. A., Folting, K. & Merrit, L. L. Jr (1969). *Acta Cryst. B* **25**, 88–93.
 Zsolnai, L. (1996). ZORTEP. Program for the Presentation of Thermal Ellipsoids. University of Heidelberg, Germany.

Acta Cryst. (1998). C **54**, 428–430

(3*S*)-4,4-Dimethyl-2-oxotetrahydrofuran-3-yl (2*S*)-2-(1,4-Benzodioxin-6-yl)propionate

M. TERESA VÁZQUEZ,^a M. DOLORS PUJOL^a AND XAVIER SOLANS^b

^aLab. Química Farmacèutica, Universitat de Barcelona, Diagonal s/n, E-08028 Barcelona, Spain, and ^bDpto. de Cristal·lografia, Mineralogia i Dipòsits Minerals, Universitat de Barcelona, Martí i Franquès s/n, E-08028 Barcelona, Spain. E-mail: xavier@natura.geo.ub.es

(Received 20 June 1997; accepted 26 September 1997)

Abstract

The 2-oxofuran moiety in the title compound, C₁₇H₁₈O₆, has a skew-envelope form. The heterocycle of the 1,4-benzodioxin-6-yl moiety has the typical half-chair form and steric hindrance between the substituents produces